



Solutions

Solutions

Solvents: substances capable of dissolving other substances

Solutes: substances that dissolve in a solvent

Solution: a homogenous mixture of uniform composition

Aqueous solutions: solutions in which water is the solvent

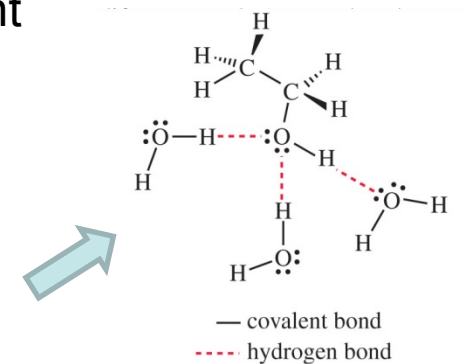
General rule: Like dissolves like

(a) Hydroxyl (OH) groups interact well with other hydroxyl groups; dissolve in water, alcohols, sugars

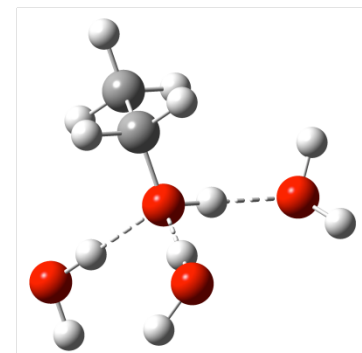
(b) Hydrocarbons (CH groups) mix well with other hydrocarbons; dissolve in oils, fats



Now that we know how water/alcohol interact, then what does a fat/oil look like?

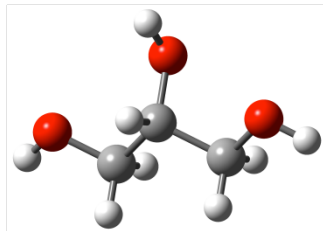
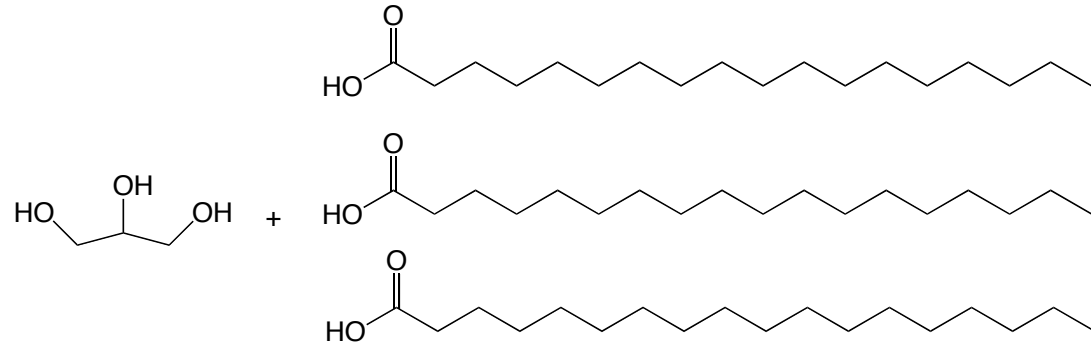


(rotatable--click on picture at least for PC)



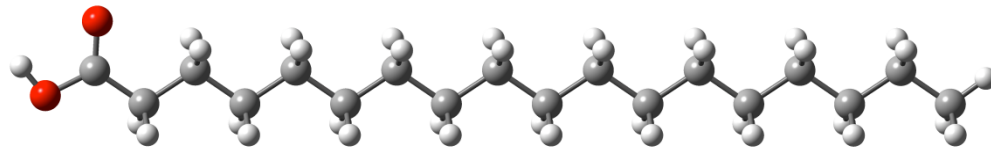
Fats, Fatty Acids, Triglycerides

Another **condensation** reaction: three **fatty acids** combine with glycerol to make a **triglyceride** and water:



glycerol

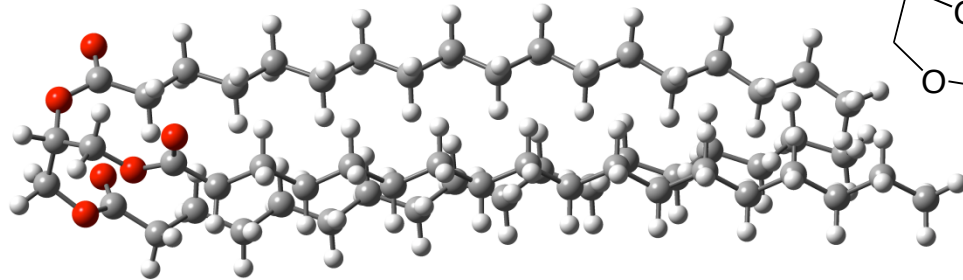
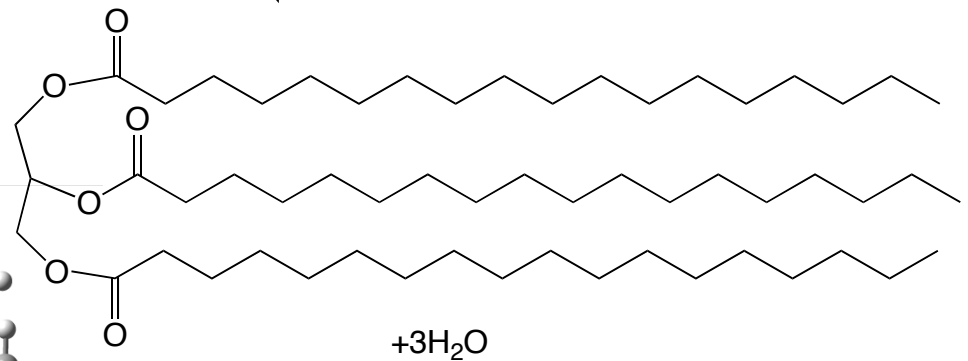
+ 3



stearic acid



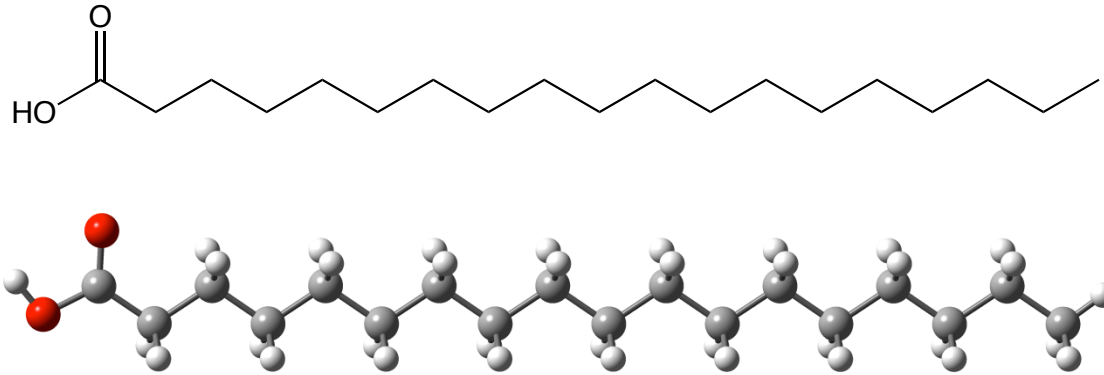
a triglyceride



Fats: Structure Impacts Physical Properties



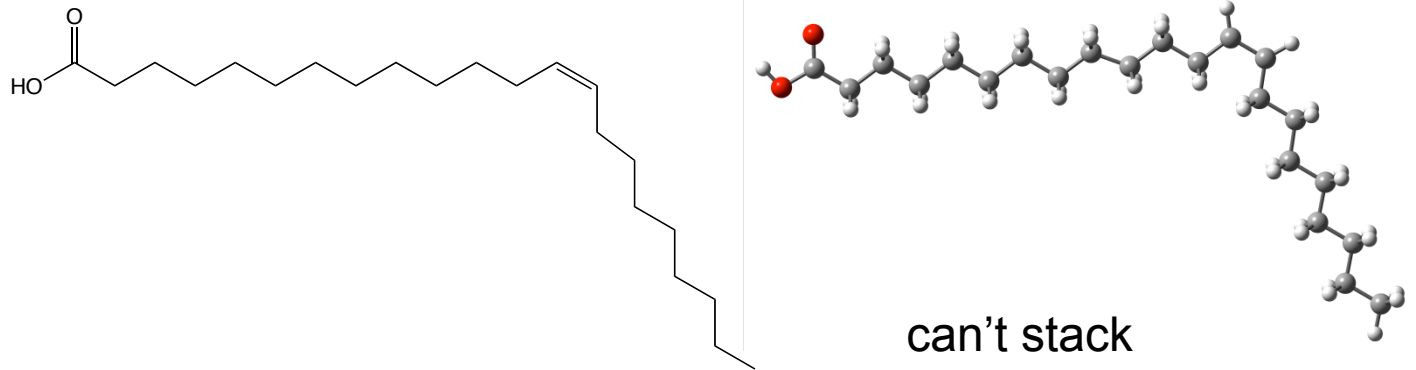
saturated fat—no double bonds in fatty acid



stackable



unsaturated fat, double bond (makes the chain bent)



can't stack

Like Dissolves Like

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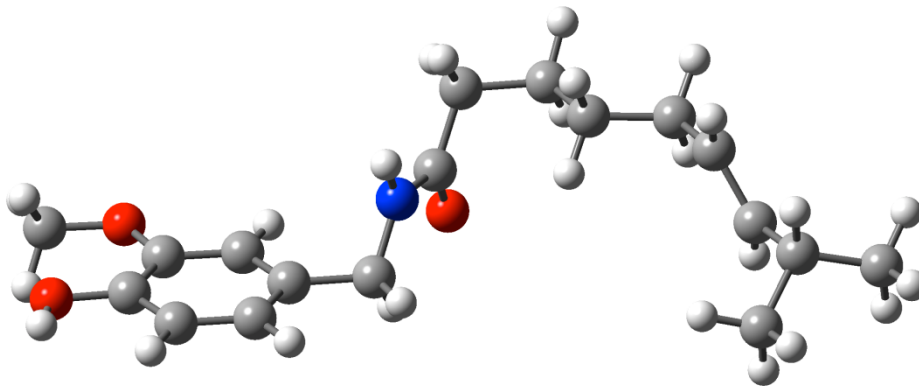
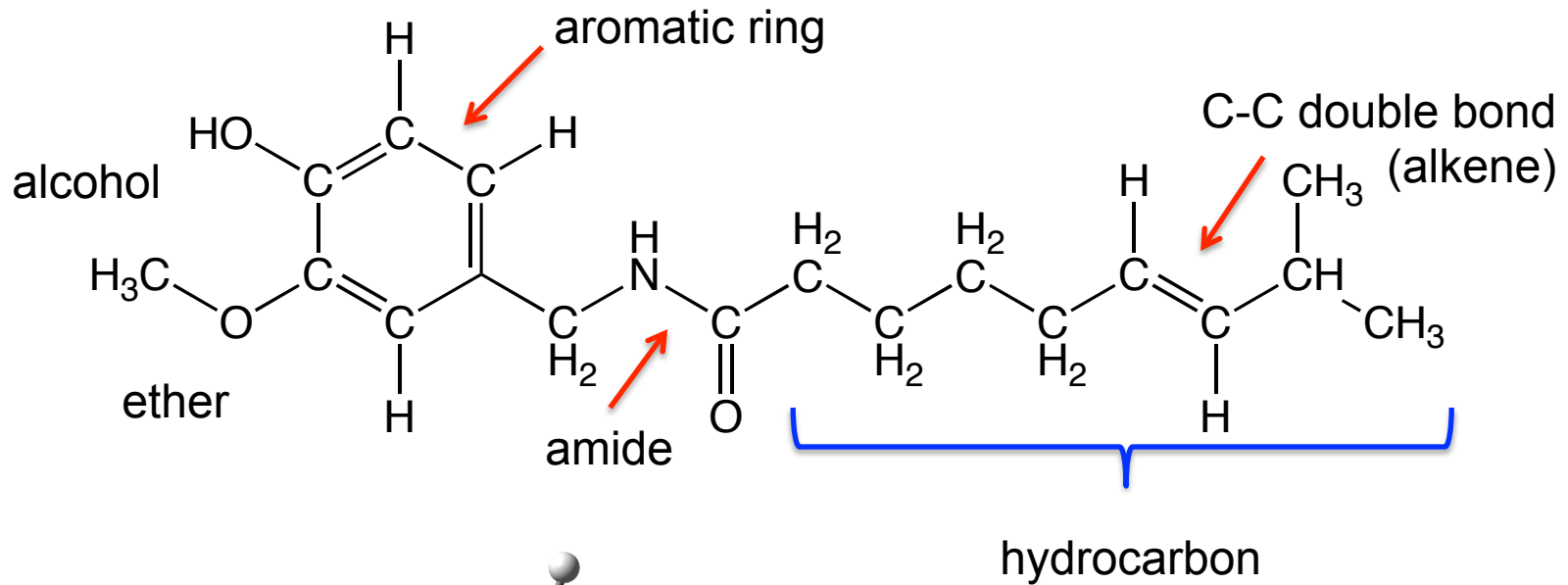
(b) Hydrocarbons (CH groups) mix well with other hydrocarbons; dissolve in oils, fats

Many molecules contain both **hydrophobic** (water repelling) and **hydrophilic** (water attracting) functional groups

Your inner (cellular) environment: bodies are ~70% water, but plenty of fat in there too (and other biologically relevant molecules)

Let's look at a situation where solubility is important...

Capsaicin



Active ingredient in chiles

Does this look more like water or fat?

Mythbusters, Season 5 Episode 23
Hot Chili Cures

Chile Heat

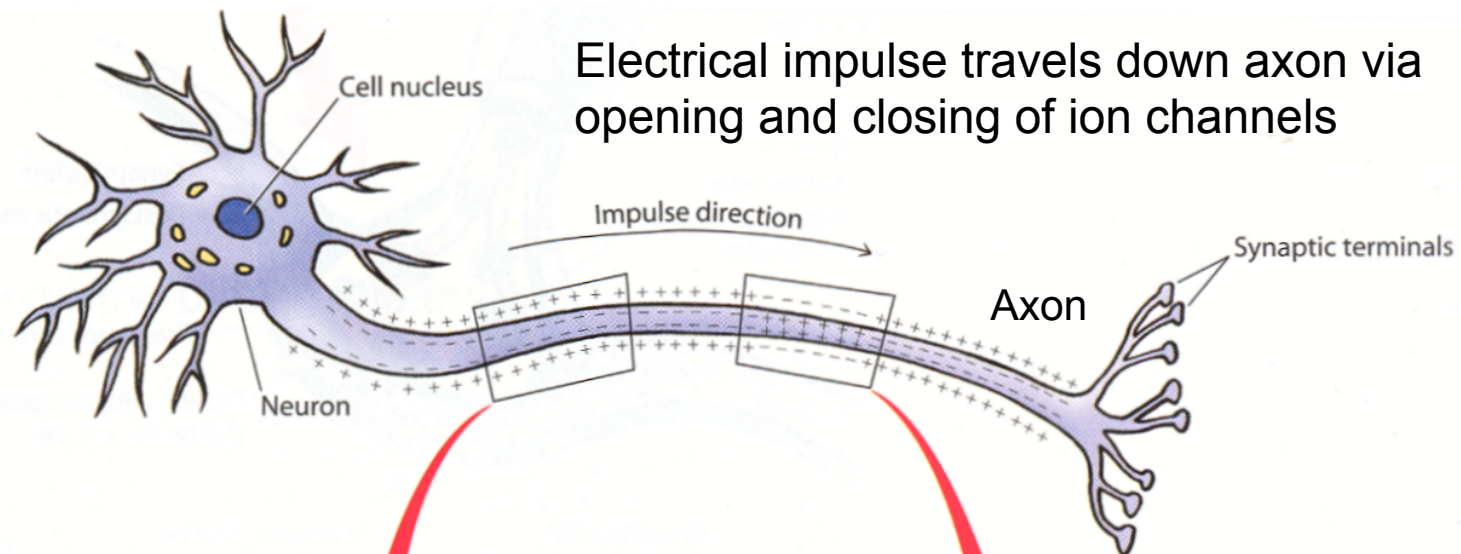


Chile Heat, Receptor Discussion



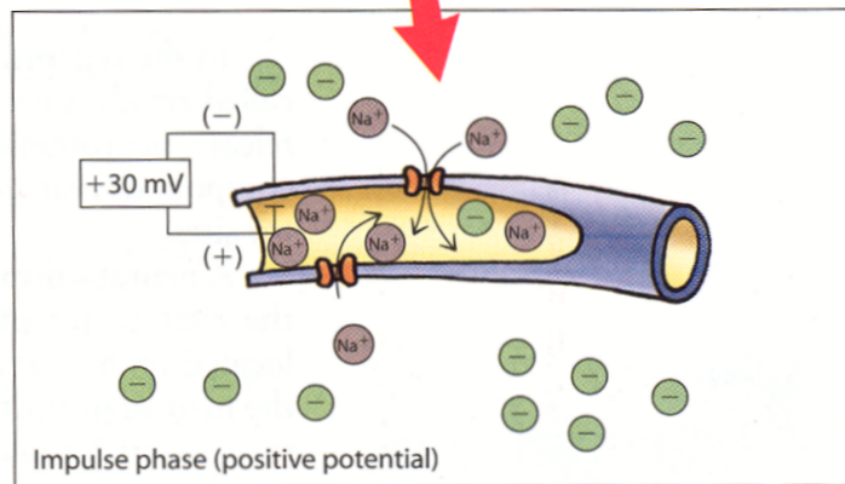
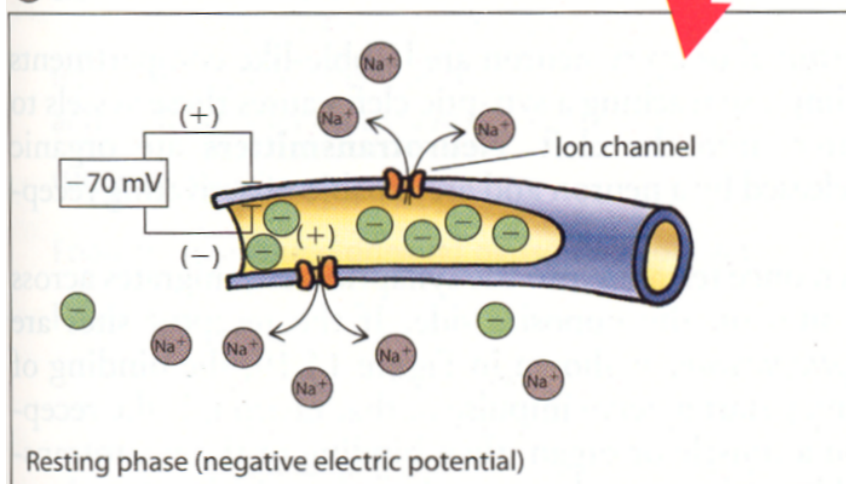
Electrochemical Transmission in Nerves

Dendrite

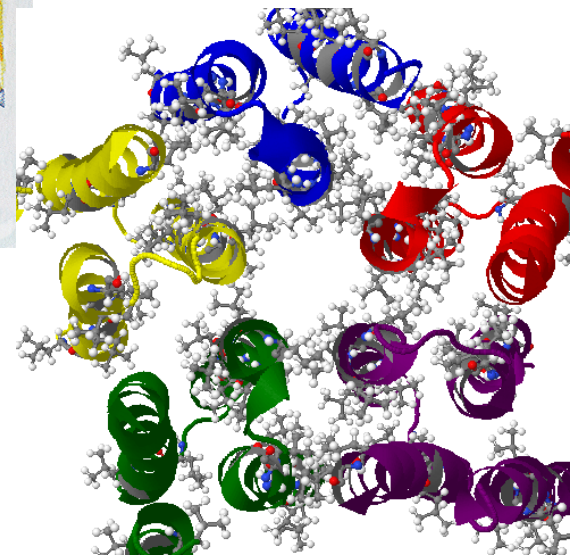
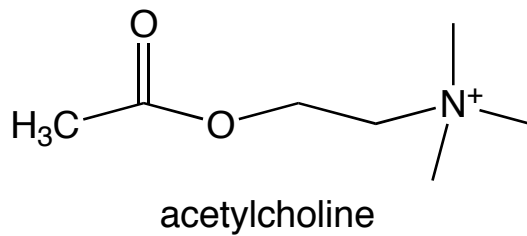
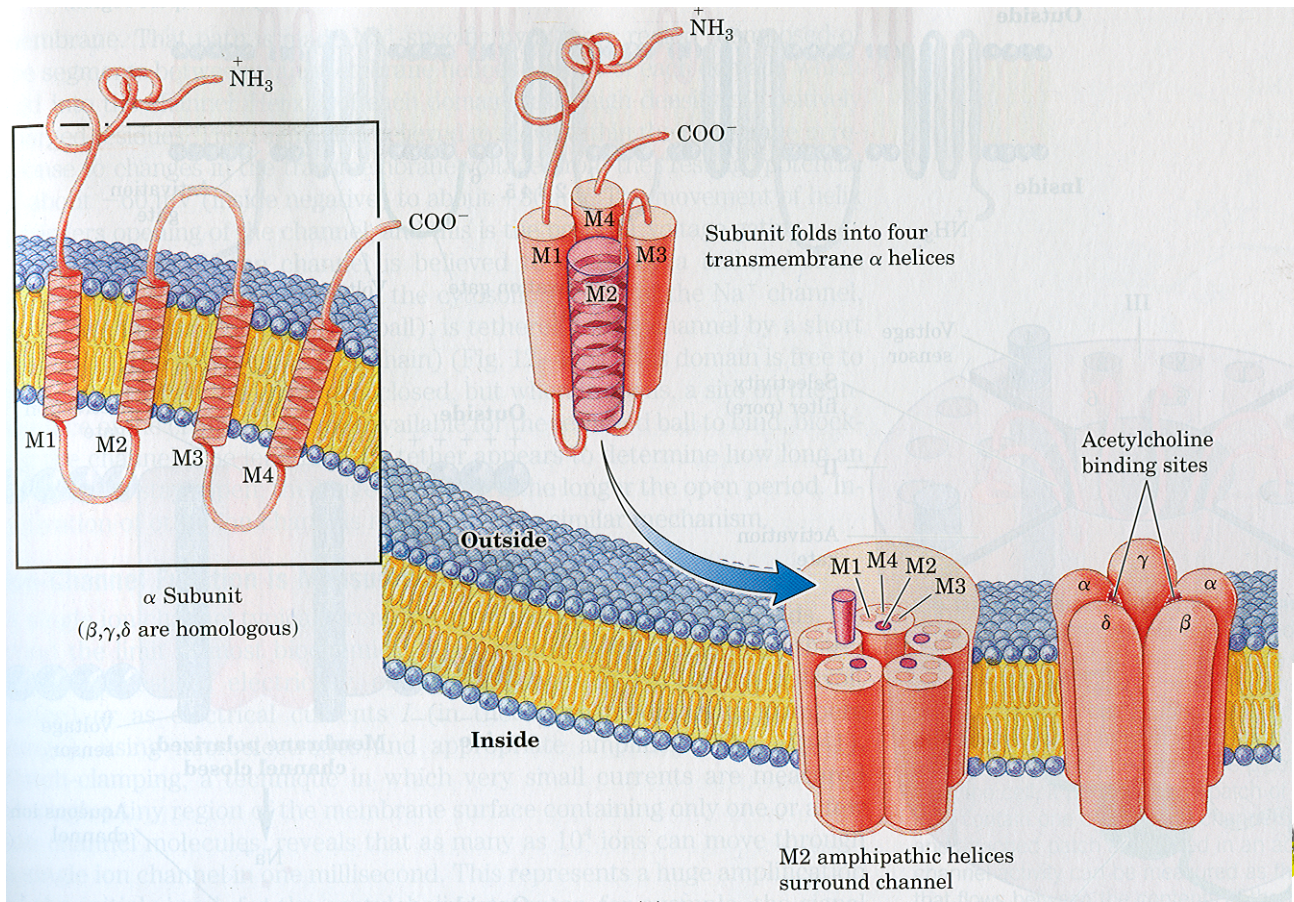


Electrical impulse travels down axon via opening and closing of ion channels

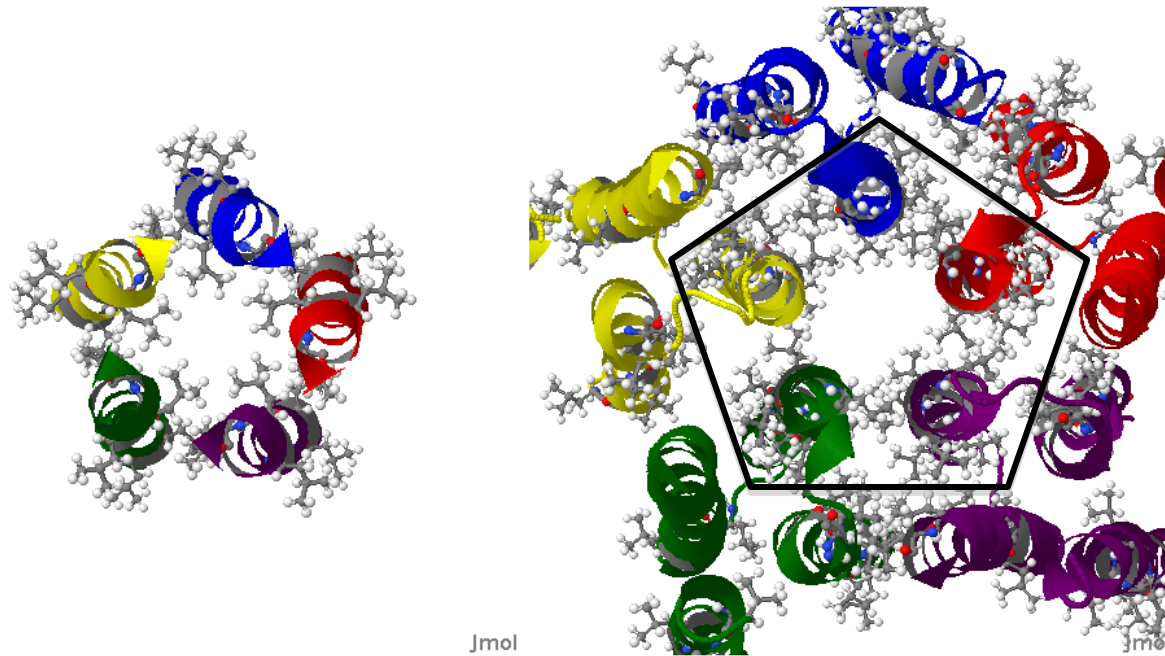
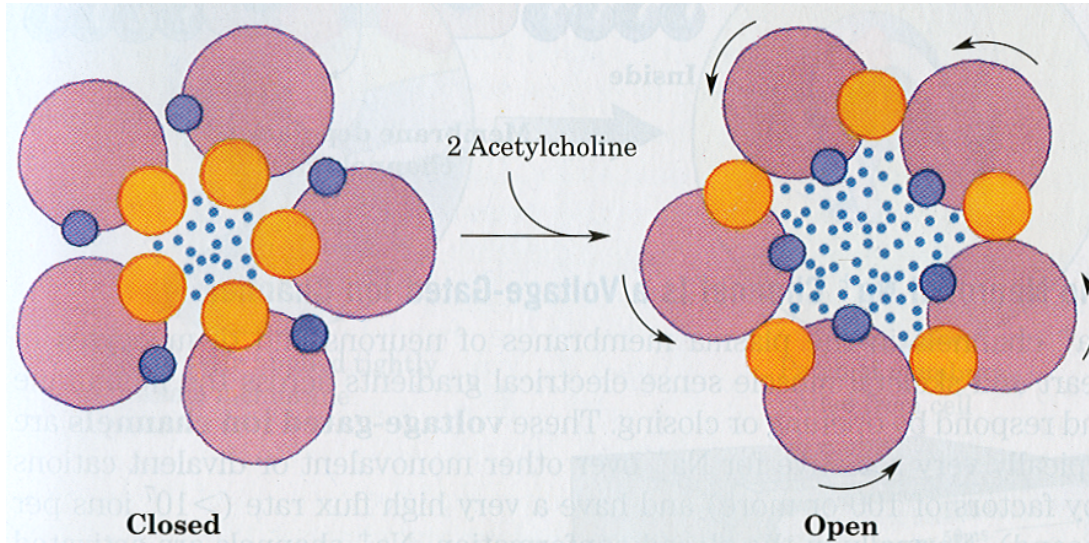
- Negative ion
- Na^+ Sodium ion



Acetylcholine channel



Channel Rotates to Open



Chile Heat: Sample Preparation



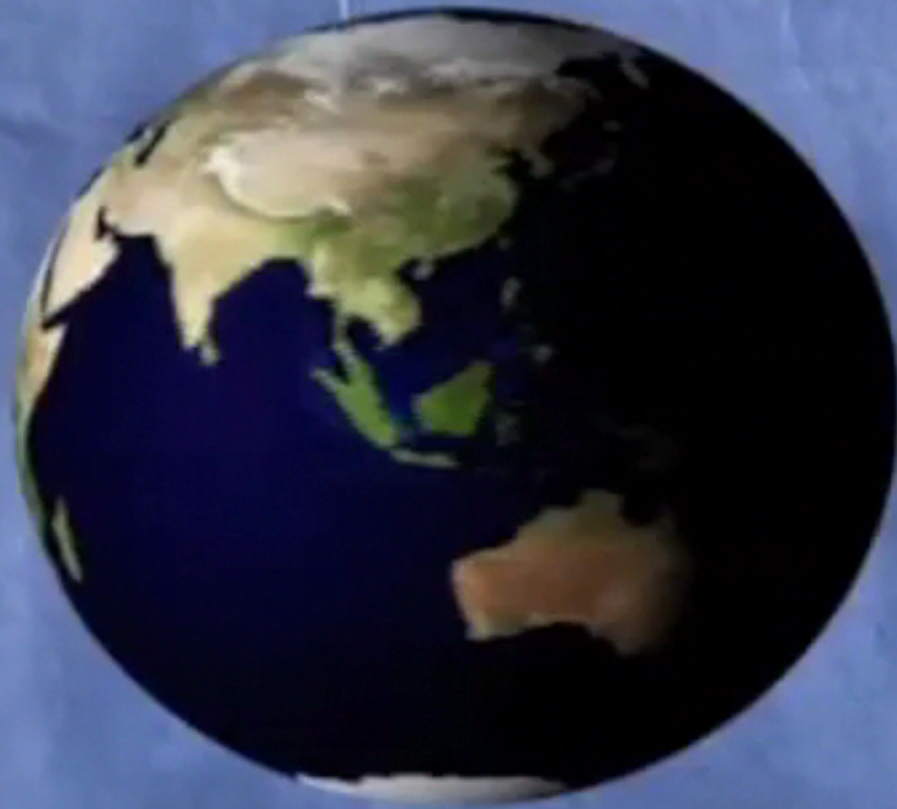
Chile Heat, more prep



Chile, First Test



Scoville

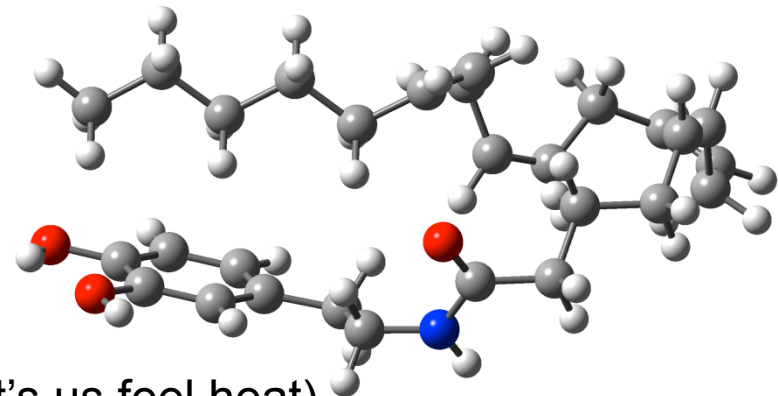
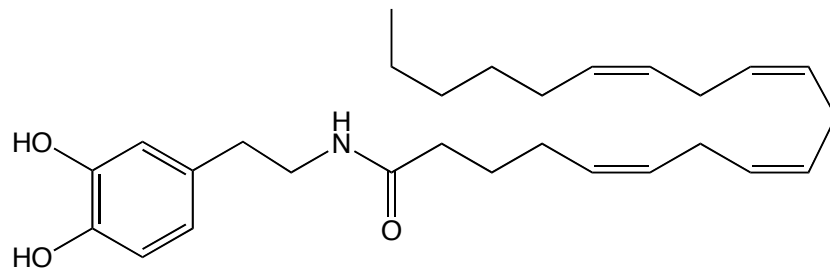
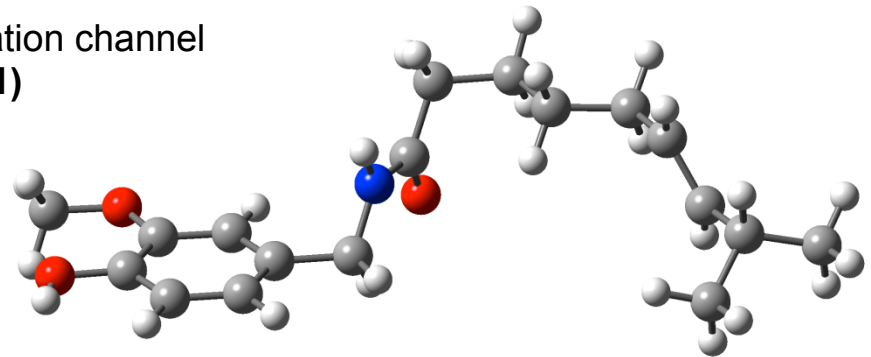
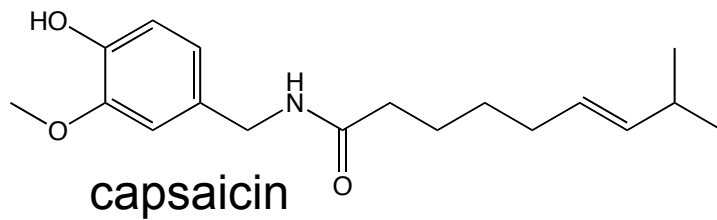
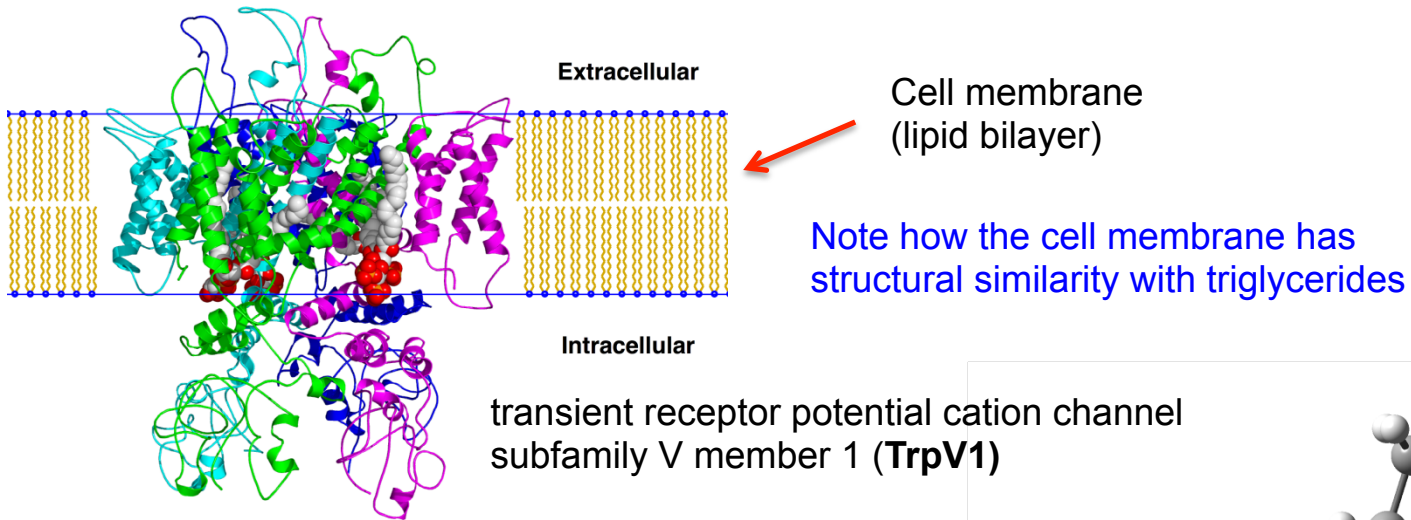


Water



Chile, other remedies

Receptor Site

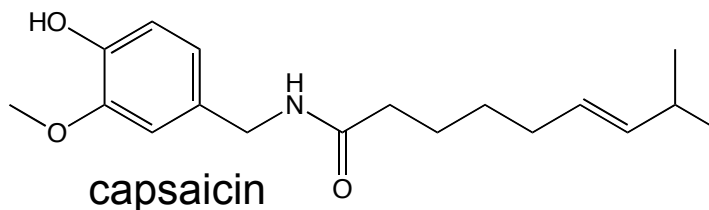


N-Arachidonoyl dopamine

Native or natural agonist (what is in us that let's us feel heat)

Recap of “Hot” Chemistry

- Capsaicin’s action and our intuitive response
 - molecule binds to the same receptor as N-Arachidonoyl dopamine—the molecule released when the cell gets too hot
 - this sets off nerve (electrical) impulses to the brain that indicate “pain”
 - thus: our body is “burning” just from eating a little fruit
 - “burning” implies that water should be thrown on the fire to put it out
- Correct response to reduce capsaicin’s burn
 - *match the molecular structure of the solute and solvent to improve solubility*
 - tested remedies:
 - milk
 - alcohol
 - toothpaste
 - petroleum jelly
 - each person has different sensitivities/responses to these small molecules, so the remedies will also be individualized



Structural Chemistry: Sweet, Fatty & Spicy

Everything is either Matter or Energy

Matter is made up of either pure substances or mixtures

Atoms have (e^- , p^+ , n)

Compounds (Molecules) built from atoms

Compounds described with Chemical formulas and structural diagrams

Compounds have 3-D structures (take up space)

Carbon forms 4 bonds (connections) Nitrogen 3, and Oxygen 2

3-D shapes of atoms connected to make 3-D molecules

Sugar

Capsaicin (chiles)

Reactions

Atoms neither created nor destroyed (balanced)

For now one important reaction: [condensation](#)

Solutions

Like dissolves like

What fats look like

What sugars look like

Functional groups we've seen:

[Alcohol/hydroxyl](#)

[Aromatic](#)

[Carboxylic acid](#)

[Ester](#)